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Key indicators

Single-crystal X-ray study

$T = 295\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$

R factor = 0.041

wR factor = 0.102

Data-to-parameter ratio = 9.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

3-Methoxysalicylaldehyde 4-methoxybenzoylhydrazone monohydrate

The title compound, $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_4 \cdot \text{H}_2\text{O}$, crystallizes as a monohydrate in which the water molecule forms bifurcated hydrogen bonds to the hydroxyl and methoxy O atoms of the methoxysalicylaldehyde portion of the Schiff base. The molecule is approximately planar; hydrogen-bonding interactions involving the amino group and the water molecule give rise to a layer structure.

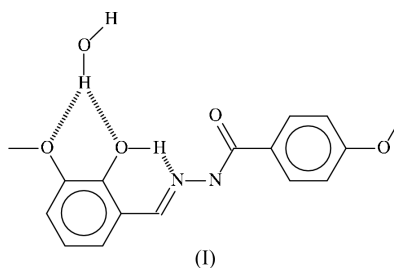
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Comment

Hydroxy-substituted benzaldehyde reagents used for condensation with benzoylhydrazine afford Schiff base hydrazones which can function as terdentate ligands towards a number of metal cations (Gao *et al.*, 1998; Liu & Gao, 1998; Chen *et al.*, 1999). In our studies on the title hydrazone, (I), we have isolated V^{5+} and Fe^{3+} complexes (Huo, Gao, Liu, Li & Ng, 2004; Huo, Gao, Liu, Li, Zhao & Zhao *et al.*, 2004; Huo, Gao, Liu, Zhao & Ng, 2004) in which the deprotonated 3-methoxysalicylaldehyde 4-methoxybenzoylhydrazone entity chelates in this manner. This hydrazone crystallizes as a monohydrate (Fig. 1).



The hydrazone is essentially flat; the aromatic ring (C10–C15) makes a dihedral angle of $5.1(5)^\circ$ with the C9/N2/O3 fragment. In the unsubstituted compound, *viz.* benzaldehyde benzoylhydrazone, the molecule is twisted with the dihedral angle between the corresponding ring and fragment being

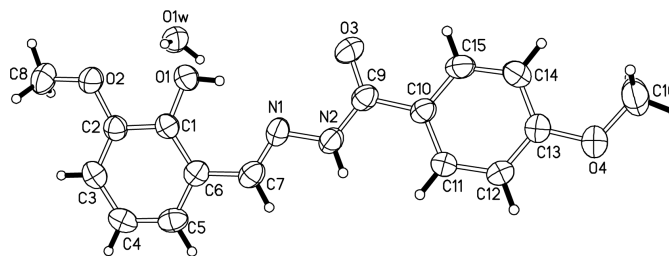


Figure 1
ORTEP (Johnson, 1976) plot of the title compound. Displacement ellipsoids for non-H atoms are drawn at the 50% probability level. H atoms are drawn as spheres of arbitrary radii.

23.6° (Litvinov *et al.*, 1991). This feature is also noted in salicylaldehyde benzoylhydrazone (dihedral angle = 21.5°; Lyubchova *et al.*, 1995). As geometry optimizations on (I) also implicate a twisted conformation, the observed planarity can be attributed to water molecules of crystallization, whose presence ensures that the molecules can be packed efficiently in a layer motif (Fig. 2). One of its H atoms engages in binding to two O atoms simultaneously (Table 2).

Experimental

An ethanol solution (25 ml) of 3-methoxysalicylaldehyde (6.04 g, 0.04 mol) was added dropwise to an ethanol solution (100 ml) of 4-methoxybenzoylhydrazine (6.44 g, 0.04 mol). The mixture was refluxed for a hour to complete the condensation. Yellow crystals (m.p. 442–444 K) were isolated from the filtered solution after a few days. Analysis calculated for C₁₆H₁₈N₂O₅: C 60.37, H 5.70, N 8.80%; found: C 60.54, H 5.95, N 8.61%.

Crystal data

C ₁₆ H ₁₆ N ₂ O ₄ ·H ₂ O	Mo K α radiation
$M_r = 318.32$	Cell parameters from 14318 reflections
Orthorhombic, $P2_12_12_1$	$\theta = 3.0\text{--}27.5^\circ$
$a = 5.013(1) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$b = 12.733(3) \text{ \AA}$	$T = 295(2) \text{ K}$
$c = 24.096(5) \text{ \AA}$	Prism, yellow
$V = 1537.9(5) \text{ \AA}^3$	$0.39 \times 0.26 \times 0.21 \text{ mm}$
$Z = 4$	
$D_x = 1.375 \text{ Mg m}^{-3}$	

Data collection

Rigaku R-Axis RAPID diffractometer	2073 independent reflections
ω scans	1784 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	$R_{\text{int}} = 0.038$
$T_{\text{min}} = 0.830$, $T_{\text{max}} = 0.979$	$\theta_{\text{max}} = 27.5^\circ$
14735 measured reflections	$h = -6 \rightarrow 5$
	$k = -16 \rightarrow 16$
	$l = -31 \rightarrow 31$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0563P)^2 + 0.2354P]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.102$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.24 \text{ e \AA}^{-3}$
2073 reflections	$\Delta\rho_{\text{min}} = -0.12 \text{ e \AA}^{-3}$
226 parameters	
H atoms treated by a mixture of independent and constrained refinement	

Table 1

Selected geometric parameters (\AA , $^\circ$).

O1—C1	1.348 (3)	N1—N2	1.382 (3)
O2—C2	1.373 (3)	N1—C7	1.284 (3)
O2—C8	1.426 (3)	N2—C9	1.353 (3)
O3—C9	1.228 (3)	C6—C7	1.458 (3)
O4—C13	1.367 (3)	C9—C10	1.490 (3)
O4—C16	1.417 (3)		
C2—O2—C8	117.2 (2)	C5—C6—C7	118.6 (2)
C13—O4—C16	118.1 (2)	N1—C7—C6	122.6 (2)
C7—N1—N2	114.5 (2)	O3—C9—N2	122.1 (2)
C9—N2—N1	120.3 (2)	O3—C9—C10	121.2 (2)
O1—C1—C2	116.8 (2)	N2—C9—C10	116.7 (2)
O1—C1—C6	123.9 (2)	C15—C10—C9	117.6 (2)
O2—C2—C1	114.3 (2)	C11—C10—C9	124.9 (2)
O2—C2—C3	125.4 (2)	O4—C13—C12	116.0 (2)
C1—C6—C7	122.2 (2)	O4—C13—C14	124.7 (2)

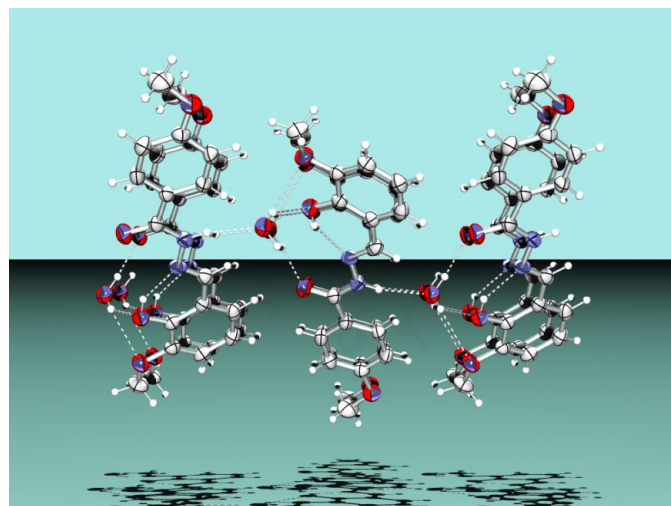


Figure 2

POV-Ray (Cason, 2002)/ORTEPII (Johnson, 1976) plot of the hydrogen-bonded layer structure, viewed along the a axis.

Table 2

Hydrogen-bonding geometry (\AA , $^\circ$).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
O1—H1 o ···N1	0.85 (1)	1.96 (2)	2.700 (2)	145 (3)
O1 w —H1 w 1···O1	0.86 (1)	2.33 (2)	3.067 (3)	145 (3)
O1 w —H1 w 1···O2	0.86 (1)	2.40 (3)	3.104 (3)	140 (4)
O1 w —H1 w 2···O3 ⁱ	0.87 (1)	1.84 (1)	2.701 (2)	171 (3)
N2—H2 n ···O1 w ⁱⁱ	0.85 (1)	2.08 (1)	2.904 (3)	165 (2)

Symmetry codes: (i) $1 + x, y, z$; (ii) $2 - x, \frac{1}{2} + y, \frac{3}{2} - z$.

The H atoms were placed in calculated positions [aromatic C—H = 0.93 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$]. Those of the methoxy groups were rotated to fit the electron density [$\text{C—H} = 0.96 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$]. The H atoms were included in the refinements in the riding-model approximation. The water and amino H atoms were located and refined with distance restraints of O—H = N—H = 0.85 (1) \AA and H···H = 1.39 (1) \AA . In the absence of significant anomalous dispersion effects, Friedel pair reflections were merged before the final refinement.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSK, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *POV-Ray* (Cason, 2002); software used to prepare material for publication: *SHELXL97*.

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